

Structural distortions in Pd–MeO-Biphep and Pd–Binap aryl complexes. Anomalies induced *via* electronic effects

Daniela Drago,^a Paul S. Pregosin,^{*a} Matthias Tschoerner^a and Alberto Albinati^{*b}

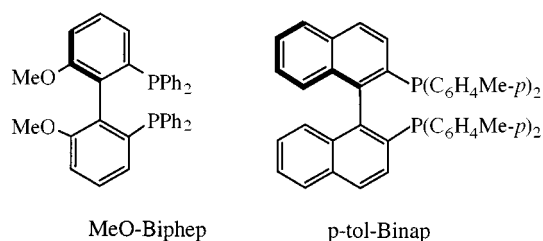
^a Laboratory of Inorganic Chemistry, ETH Zentrum, 8092 Zürich, Switzerland

^b Chemical Pharmacy, University of Milan, I-20131 Milan, Italy

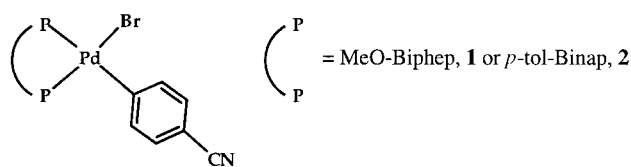
Received 4th May 1999, Accepted 1st June 1999

The four-coordinate complexes [PdBr(*p*-NCC₆H₄)-(MeO-Biphep)], **1** [MeO-Biphep = 6,6'-dimethoxy-2,2'-bis(diphenylphosphino)-1,1'-biphenyl], and [PdBr(*p*-NC₆H₄)(*p*-tol-Binap)], **2** [*p*-tol-Binap = 2,2'-bis(di-*p*-tolylphosphino)-1,1'-binaphthyl], both distort markedly, with the former no longer square-planar, and the latter, possessing the longest recorded Pd–P bond length.

The atropisomeric chiral bidentates MeO-Biphep and Binap are



recognised to be excellent auxiliaries in a number of enantioselective transformations,^{1–3} including C–C coupling.⁴ Pd–aryl complexes, which are intermediates in both cross-coupling and Heck chemistry,⁴ arise from the oxidative addition of aryl–X compounds, X = I, Br, OTs, to Pd(0) complexes.^{5–10} In connection with kinetic studies related to the enantioselective Heck reaction we recently noted¹¹ that aryl intermediates of the type [PdX(aryl)(MeO-Biphep)], with electron-donor groups on the aryl, are not very stable. We report here that the structures of the compounds [PdBr(*p*-NCC₆H₄)(MeO-Biphep)], **1**, and [PdBr(*p*-NCC₆H₄)(*p*-tol-Binap)], **2**, which are more stable, both distort markedly (but very differently) in order to accommodate the electronic pressure exerted by the aryl groups.†



The molecular structures for **1** and **2** were determined *via* X-ray diffraction methods‡ and ORTEP views of these molecules are given in Figs. 1 and 2. Selected bond distances and bond angles are given in the captions. In compound **1** there is sufficient space for the aryl and bromide ligands; nevertheless both ligands deviate so strongly from the coordination plane defined by the two P-donors and the metal, +0.88 Å and –0.57 Å, respectively, that one can no longer speak of a square planar geometry. We envision the structure as arising *via* rotation of the P–Pd–P and Cl–Pd–Br, planes, relative to one another, and not due to a tetrahedral distortion (note that the P–Pd–P and C1L–Pd–Br angles are *ca.* 94° and 90°, respectively). The Pd–C1L separation of 2.104(3) Å is relatively long [2.04–2.06 Å is normal for MeO-Biphep¹¹ although, in general much shorter Pd–C(aryl) distances have been reported^{12,13}]. The ring of the

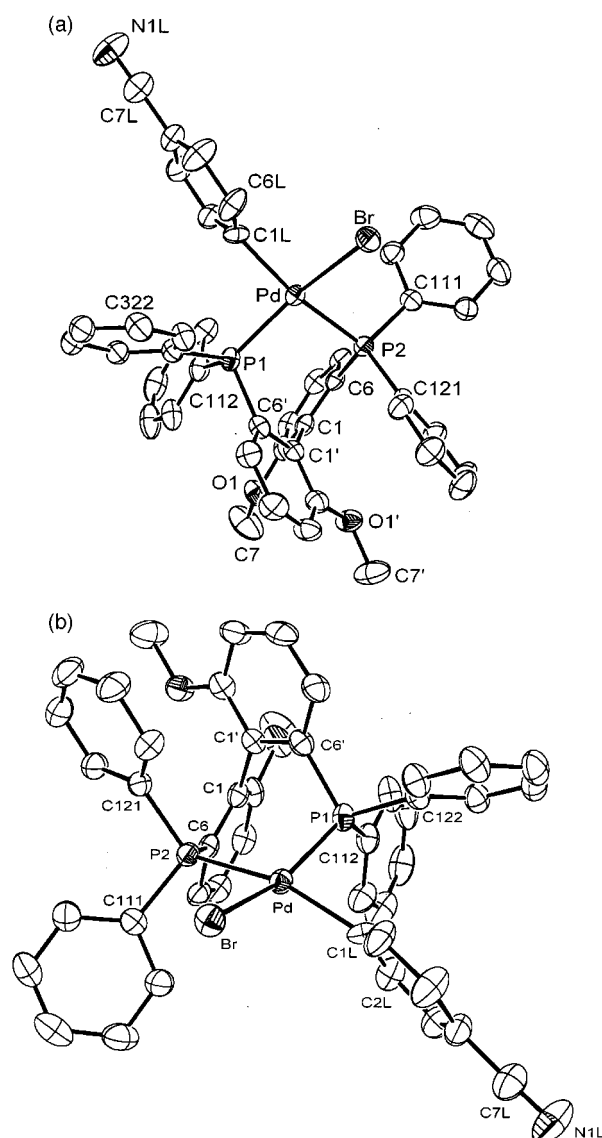


Fig. 1 ORTEP²⁰ views of complex **1**. (a) From above and (b) from behind (and slightly above) the *p*-cyanoaryl ligand. Selected bond lengths (Å) and bond angles (°) for **1**: Pd–P1 2.2700(9), Pd–P2 2.3501(9), Pd–C1L 2.104(3), Pd–Br 2.4920(4), C7L–N1 1.126(5); P1–Pd–P2 94.06(3), P2–Pd–C1L 163.02(10), P1–Pd–C1L 92.22(9), P1–Pd–Br 158.96(3), P2–Pd–Br 89.69(2), C1L–Pd–Br 90.09(9).

p-NCC₆H₄ ligand makes an angle of *ca.* 73° with the P–Pd–P plane, but seems to be somewhat bent, P2–Pd–C1L = 163°, see Fig. 1a. We find no evidence for strain due to packing effects.

For the Binap complex **2** the observed coordination geometry is slightly distorted square planar, see Fig. 2. Relative to **1** the Pd–C1L bond separation is now 0.078 Å shorter, 2.026(6) Å,¹⁴ however, a search of the Cambridge database reveals that the

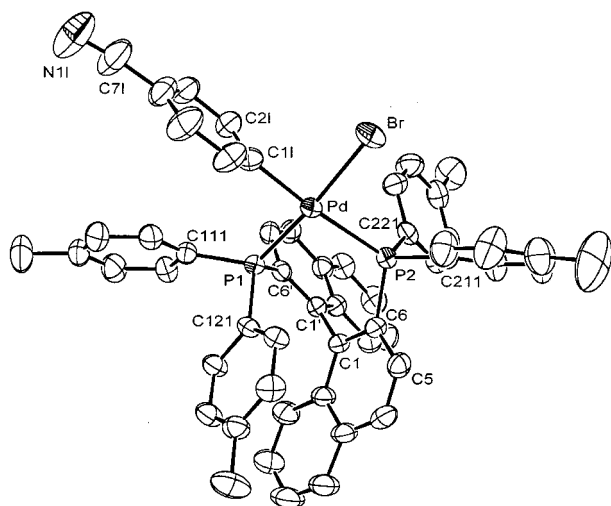
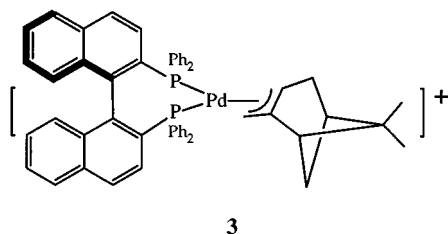


Fig. 2 ORTEP view of complex **2** looking down on the *p*-cyanoaryl ligand and the coordination plane. Selected bond lengths (Å) and bond angles (°) for **2**: Pd–P1 2.254(1), Pd–P2 2.437(1), Pd–C1L, 2.026(6), Pd–Br 2.4484(7), C7–N1 1.12(1); P1–Pd–P2 92.38(5), P2–Pd–C1L 175.6(2), P1–Pd–C1L 91.0(2), P1–Pd–Br 173.51(4), P2–Pd–Br 90.90(4), C1L–Pd–Br 85.4(2).

Pd–P2 bond, *trans* to the aryl, at 2.437(1) Å represents the longest Pd–P bond ever reported. § For comparison, in the Binap β-pinene allyl complex [Pd(η³-C₁₀H₁₅){(*R*)-(+)-Binap}]·[CF₃SO₃], **3**, the Pd–P distances are 2.312(3) Å and 2.347(5)



Å.¹⁵ The observed Pd–P2 bond in **1**, 2.3501(9) Å, is much shorter, but lies towards the upper end of the literature range (Pd–P separations of the order of *ca.* 2.20–2.36 Å are common^{16–19}). In both **1** and **2** the *p*-CN group appears to be a normal triple bond (see captions).

Despite the superficial similarity of the ligands, *i.e.*, biaryl-based triaryl phosphine types, the structures for **1** and **2** are very different. Whereas the extreme lengthening of the Pd–P2 bond in the Binap complex **2** arises due to good donor properties of the aryl group, the MeO-Biphep analog **1** avoids this electronic strain by strongly deviating from square planar geometry and thus weakening the Pd–C1L overlap. Given the rather novel structural results for these *p*-NCC₆H₄ aryl compounds, it is not surprising that we cannot readily isolate analogous complexes with aryl ligands which are even more electron donating, *e.g.* *p*-MeC₆H₄ and *p*-MeOC₆H₄.

Acknowledgements

P. S. P. thanks the Swiss National Science Foundation, and the ETH Zurich for financial support. A. A. thanks MURST and the Vigoni Foundation for support. We also thank Johnson Matthey for the loan of PdCl₂ and F. Hoffmann La Roche, Basel, for the MeO-Biphep ligand.

Notes and references

† The complexes were prepared as described in reference 11.

‡ Crystal data for compound **1**: C₄₅H₃₆BrNO₂P₂Pd, *M* = 871.00, orthorhombic, space group *P*2₁2₁2₁ (no. 18), *a* = 11.4485(2), *b* = 17.3604(1), *c* = 19.0940(3) Å, *U* = 3794.94(9) Å³, *Z* = 4, *μ* = 16.66 cm⁻¹, *T* = 200 K, *R*₁ = 0.0267 (for 5794 unique reflections having *I* > 2σ(*I*)), 0.0354 (for all 6492 independent reflections). Crystal data for compound **2**·CH₂Cl₂: C₅₆H₄₆BrCl₂NP₂Pd, *M* = 1052.16, monoclinic, space group *C*2 (no. 5), *a* = 29.0943(4), *b* = 11.8671(2), *c* = 17.2953(2) Å, β = 116.790(3)°, *U* = 5330.51(13) Å³, *Z* = 4, *μ* = 12.94 cm⁻¹, *T* = 293 K, *R*₁ = 0.0419 (for 5550 unique reflections with *I* > 2σ(*I*)), 0.0494 (for all 7049 independent reflections). CCDC reference number 186/1478.

§ Note added at proof: a Pd–P bond length of *ca.* 2.5 Å (*trans* to SiCl₃) has been observed.²¹

- R. Noyori, *Chimia*, 1988, **42**, 215; R. Noyori, *Asymmetric Catalysis in Organic Synthesis*, John Wiley and Sons, New York, 1994.
- Y. Cramer, J. Foricher, U. Hengartner, C. Jenny, F. Kienzle, H. Ramuz, M. Scalone, M. Schlageter, R. Schmid and S. Wang, *Chimia*, 1997, **51**, 303; R. Schmid, M. Cereghetti, B. Heiser, P. Schönholzer and H. J. Hansen, *Helv. Chim. Acta*, 1988, **71**, 897; R. Schmid, E. A. Broger, M. Cereghetti, Y. Cramer, J. Foricher, M. Lalonde, R. K. Mueller, M. Scalone, G. Schoettel and U. Zutter, *Pure Appl. Chem.*, 1996, **68**, 131; C. Bolm, D. Kaufmann, S. Gessler and K. Harms, *J. Organomet. Chem.*, 1995, **502**, 47.
- G. Trabesinger, A. Albinati, N. Feiken, R. W. Kunz, P. S. Pregosin and M. Tschoerner, *J. Am. Chem. Soc.*, 1997, **119**, 6315.
- L. E. Overman and J. T. Link, in *Metal Catalyzed Cross-Coupling Reactions*, ed. F. A. S. Diederich, P. J. Weinheim, 1998.
- A. de Meijere and F. E. Meyer, *Angew. Chem.*, 1994, **106**, 2473; C. Amatore, E. Carré, A. Jutand, M. A. M'Barki and G. Meyer, *Organometallics*, 1995, **14**, 5605.
- P. Garrou and R. F. Heck, *J. Am. Chem. Soc.*, 1976, **98**, 4115; R. F. Heck, *Acc. Chem. Res.*, 1979, **12**, 146; R. F. Heck, *Comprehensive Organic Synthesis*, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1991.
- O. Loiseleur, P. Meier and A. Pfaltz, *Angew. Chem.*, 1996, **108**, 218.
- F. Ozawa, A. Kubo and T. Hayashi, *J. Am. Chem. Soc.*, 1991, **113**, 1417.
- M. Tschoerner, G. Trabesinger, A. Albinati and P. S. Pregosin, *Organometallics*, 1997, **16**, 3447.
- J. F. Hartwig, *Angew. Chem., Int. Ed.*, 1998, **37**, 2090.
- M. Tschoerner, P. S. Pregosin and A. Albinati, *Organometallics*, 1999, **18**, 670.
- A. G. Orpen, L. Brammer, F. H. Allen, O. Kennard, D. G. Watson and R. Taylor, *J. Chem. Soc., Dalton Trans.*, 1989, S1; W. J. Marshall, D. L. Thorn and V. V. Grushin, *Organometallics*, 1998, **17**, 5427.
- J. M. Brown, J. Perez-Torrente, N. Alcock and H. J. Clase, *Organometallics*, 1995, **14**, 207. The Pd–C distance in the aryl complex PdI{[2-(CH₂OCH=CH₂)C₆H₄](dppf)} is 2.055(7) Å.
- J. M. Brown and J. J. P. Torrente, *Organometallics*, 1995, **14**, 1195. Pt–C separations of 2.090(31) Å and 2.084(40) Å are reported for a biaryl-Binap complex. The Pt–P bond distances are 2.303(8) Å and 2.301(8) Å.
- P. S. Pregosin, H. Rügger, R. Salzmann, A. Albinati, F. Lianza and R. W. Kunz, *Organometallics*, 1994, **13**, 83.
- J. M. Wisner, T. J. Bartczak and J. A. Ibers, *Organometallics*, 1986, **5**, 2044.
- W. A. Herrmann, C. Brossmer, T. Priermeier and K. Oefele, *J. Organomet. Chem.*, 1994, **481**, 97.
- G. Mann, D. Baranano, J. F. Hartwig, A. L. Rheingold and I. A. Guzei, *J. Am. Chem. Soc.*, 1998, **120**, 9205.
- D. K. Wicht, M. A. Zhuravel, R. V. Gregush, D. S. Glueck, I. A. Guzei, L. M. Liable-Sands and A. L. Rheingold, *Organometallics*, 1998, **17**, 1412.
- C. K. Johnson, ORTEP, Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, TN, 1976.
- A. Togni, personal communication.

Communication 9/03509E